

PH. 1A PHYSICS

REVISION NOTES

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Errors

Mechanics (classical)

Kinetic Theory

Intermolecular forces

Oscillations

Waves

Fields

Special Relativity

Quantum Mechanics

Measurements, errors, units

<u>The SI set</u>	mass [M]	-	Kilogram	/kg
	length [L]	-	metre	/m
	time [T]	-	second	/s
	electric charge	-		
	flow - current [I]	-	Amp	/A
	Amount of substance	-	mole	/mol
	Thermodynamic temperature	-	Kelvin	/K
	luminous intensity	-	candela	/cd

Fundamental units used for dimensional analysis (check physical expressions are homogeneous)

Error analysis

Sample mean $\bar{x} = \frac{1}{n} \sum_{i=1}^n x_i$ Standard deviation $s = \frac{1}{\sqrt{n}} \sqrt{\sum_{i=1}^n (x_i - \bar{x})^2}$

Standard deviation of a population $\sigma = \sqrt{\frac{n}{n-1}}$

Standard deviation of means $\sigma_m = \frac{\sigma}{\sqrt{n}}$

⇒ quote measurement $\bar{x} \pm \sigma_m$

error propagation formula (in a product of independent [random] variables)

If $z = z(A, B, C, D, \dots)$ then error in z σ_z is:

$$(\sigma_z)^2 = \left(\frac{\partial z}{\partial A} \sigma_A \right)^2 + \left(\frac{\partial z}{\partial B} \sigma_B \right)^2 + \dots = \sum_i \left(\frac{\partial z}{\partial x_i} \sigma_i \right)^2$$

Now if errors are quoted as fractions (or proportions)

$$\text{If } z = \prod_i x_i^{k_i} \Rightarrow \left(\frac{\sigma_z}{z} \right)^2 = \sum_i (k_i p_i)^2$$

where p_i is the fractional error in x_i .

For a sum of products i.e. $z = z_1 + z_2$
we find σ_z as $\sigma_{z_1} + \sigma_{z_2}$.

Classical Mechanics

Newton's laws of motion

I Every body continues in uniform motion in a straight line unless it is compelled to change by an action of a force

II $\underline{F} = m\underline{a}$ or more generally $\underline{F} = d\underline{p}/dt$
 $\underline{F} = \underline{p}_{\text{net}}$ $\underline{p} = \text{momentum, } m\underline{v}$. (NOTE directions)

III For every force there is always an equal and opposite reaction on the body which is imposing the said force.

$$\underline{F}_A(B) = -\underline{F}_B(A)$$

Dynamics of a body

(1) Linear dynamics A force \underline{F} will act according to Newton II on the centre of mass of a rigid body. (Reducing a body to a 'particle' with all the mass of the body).

(2) Rotational dynamics A force applied away from the centre of mass of the body will exert a torque as well as a linear acceleration of the C.O.M. Torque is the rate of change of angular momentum.

$$\underline{L} = I\underline{\omega} = \sum_i (\underline{r}_i \wedge \underline{p}_i)$$

(\underline{L} = angular momentum, $\underline{\omega}$ = angular velocity, I = moment of inertia, \underline{p} = momentum (linear) \underline{r} = vector \perp to axis or rotation through C.O.M. to applied force vector)

$$I = \sum_i m_i |\underline{r}_i|^2 \quad \underline{\omega} = \frac{d\theta}{dt} \quad \underline{\omega} \parallel \underline{L}$$

$$\text{Torque} = \underline{\tau} = \underline{r} \wedge \underline{F} = \frac{d}{dt} \underline{L} = \frac{d}{dt} (I\underline{\omega})$$

or
velocity
vector.

Centre of mass

For an arbitrary axis and same point of \underline{r} as before:

$$\underline{\bar{r}} \wedge m_{\text{tot}} \underline{g} = \sum_i \underline{r}_i \wedge m_i \underline{g}$$

defines centre of gravity $\underline{\bar{r}}$ of body composed of mass elements m_i . In a uniform \underline{g} field this = centre of mass

Work, energy, impulse

linear work on C.O.M

$$W_L = \int_r \underline{F} \cdot d\underline{r} \quad d\underline{r} = \text{element of path } r$$

rotational work on body

$$W_R = \int_{\theta} \underline{G} \cdot d\underline{\theta} \quad d\underline{\theta} = \text{angular displacement about axis}$$

NOTE / $\int \underline{F} \cdot d\underline{r} = \frac{1}{2} m v^2 - \frac{1}{2} m u^2$ $u = \text{initial speed of COM}$
 $v = \text{final " " "}$

$$\int \underline{G} \cdot d\underline{\theta} = \frac{1}{2} I \omega^2 - \frac{1}{2} I \Omega^2 \quad \Omega = \text{initial angular speed of body}$$

$\omega = \text{final " " "}$
" " "

change of momentum = impulse

linear impulse

$$\int \underline{F} dt = \Delta \underline{p}$$

rot. impulse

$$\int \underline{G} dt = \Delta \underline{L}$$

laws of conservation

Mass, energy, charge and momentum are all conserved in an isolated system.

Note energy can be stored in 'potential wells' \rightarrow fields.

$$\frac{1}{2} m v_i^2 + \phi_i = \frac{1}{2} m v_f^2 + \phi_f$$

where $\frac{1}{2} m v^2 = \text{kinetic energy}$ $\phi = \text{potential energy}$.

Frames of reference

Coordinate systems used for quantities describing events.

use pair vectors. $[r, t]$

see "Special Relativity" notes for standard configuration.

In classical mechanics one transforms between inertial frames (where Newton I holds \rightarrow i.e. no pres (net)) using Galilean relativity.

$$x' = x - vt$$

$$z' = z$$

$$y' = y$$

$$t' = t$$

NOTE invariant nature of time.

NOTE a useful frame of reference for collision problems is the zero momentum frame:

\rightarrow we add a velocity \underline{v} to all particles

such that $\sum_i p_i = 0$.

This \Rightarrow

$$\underline{v} = - \frac{\sum_i m_i \underline{u}_i}{\sum_i m_i}$$

centre of percussion

For a vertically suspended body: $x = \frac{I}{mh}$

x = distance below C.O.M. of C.O. percussion. h = distance / height from C.O.M. to pivot.

Gyroscope

$$\text{Precession rate } \Omega = \frac{mgl}{I\omega}$$

Kinetic Theory

Main Assumptions

- ① large numbers of molecules - dynamics* can be treated statistically.
- ② Molecules are in continuous and rapid motion. (Random) collide elastically
- ③ Pressure originates from the summation of large numbers of reaction pres as molecules bounce off the body which is feeling the pressure.

* and kinematics!

For an IDEAL GAS we can make further assumptions

- (1) molecules occupy a negligible fraction of total volume.
- (2) long range forces of attraction between particles are negligible.

⇒ Ideal gas law: or $PV = nRT$
 $PV = NkT$

(P = pressure, V = volume, n = no. of moles, R = molar gas constant, N = no. of molecules, k = Boltzmann constant, T = absolute temp.)

Probability of molecules

arriving in a direction $\theta \rightarrow \theta + d\theta$
(in 2D polar's)

$$d\theta P(\theta) = \frac{1}{2} \sin\theta d\theta$$

↑ probability density function.

* Assume spherical symmetry.
→ applies 3D.

Molecular flux on a surface within an ideal gas

$$N = \frac{1}{4} n \langle c \rangle$$

(n = number density, $\langle c \rangle$ = mean molecular speed).

Microscopic derivation of pressure

$$P = \frac{1}{3} nm \langle c^2 \rangle$$

Equipartition of energy (classical!)

Each quadratic term in the energy expression contributes $\frac{1}{2}kT$ to the average energy \bar{E} .

→ must regard statistical importance of quadratic energy term to assess how \bar{E} relates to the \bar{E} averaged over a bulk mass of average temp T .

Mean free path (λ)

$$\lambda = \frac{1}{\sqrt{2} n s^2}$$

(s = molecular diameter)

Survival probability

$$P(x) dx = e^{-x/\lambda}$$

↑ probability density

(x = distance travelled before collision).

Random walks

$$\text{random walk} = \sqrt{N}$$

(N = no. of 'steps').

average

$$\frac{\langle c^2 \rangle}{c}$$

Note, in time t molecule makes on average $\frac{\langle c^2 \rangle}{c}$ steps / collisions. So random walk can be written:

$$\text{random walk} = \sqrt{t \langle c^2 \rangle}$$

Steps
= collisions.

Diffusion

$$J_z = -\frac{1}{3} \langle c \rangle \frac{dn}{dz}$$

mean free path.

(J_z = flux of gas in 'z' direction, $\frac{dn}{dz}$ = concentration gradient in z direction).

Thermal conductivity

$$Q_z = -\frac{1}{3} \rho \langle c \rangle c_v \frac{dT}{dz}$$

(ρ = density, c_v = heat capacity, Q = heat)

Viscosity

$$f_{xz} = \frac{1}{3} \rho \langle c \rangle \left(A \frac{dv_x}{dz} \right)$$

(f_{xz} = ~~momentum~~ pressure on plate of gas as a plate moves with velocity v_x a distance z above it, A = plate area)

Boltzmann Distribution

$$n_i = e^{-\epsilon_i/kT}$$

$$\Rightarrow P(\epsilon_i) = \frac{\exp(-\epsilon_i/kT)}{\sum_i \exp(-\epsilon_i/kT)}$$

ns *n_tot*

n_i = number of particles in energy state ϵ_i (of energy ϵ_i)

$P(\epsilon_i)$ = probability of finding a particle of energy ϵ_i

→ example: isothermal atmosphere: $P = P_0 \exp\left(-\frac{mgh}{kT}\right)$
(use of Boltzmann factor $\exp(-\epsilon/kT)$).

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