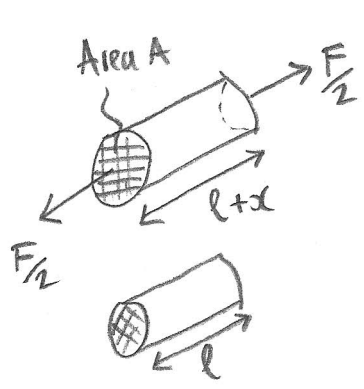


MATERIALS



Stress $\sigma = \frac{force}{area}$ i.e. $\sigma = \frac{F}{A}$

Strain ϵ is $\frac{extension(x)}{original\ length(l)}$ $\epsilon = \frac{x}{l}$

Young's modulus $Y = \frac{\sigma}{\epsilon} \Rightarrow \sigma = \epsilon Y$

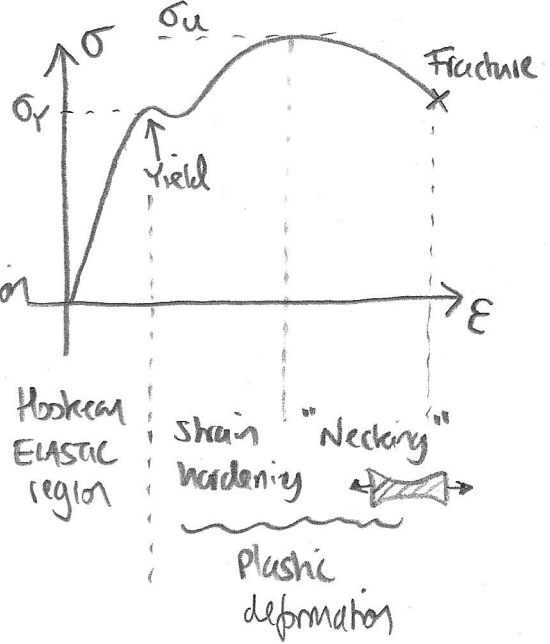
If a material is elastic and obeys **Hooke's law** $F = kx$

Elastic strain energy $E = \frac{1}{2} kx^2$

Strain energy density $U = \frac{E}{Al}$ (original volume) $= \frac{1}{2} \left(\frac{kx}{A} \right) \times \left(\frac{x}{l} \right) = \frac{1}{2} \sigma \epsilon$

$U = \frac{1}{2} Y \epsilon^2$

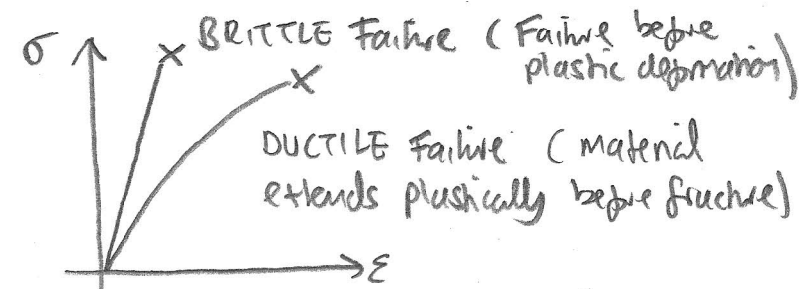
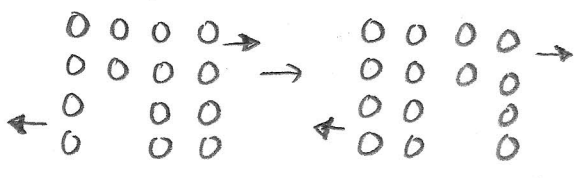
Most materials follow the **stress-strain curve**



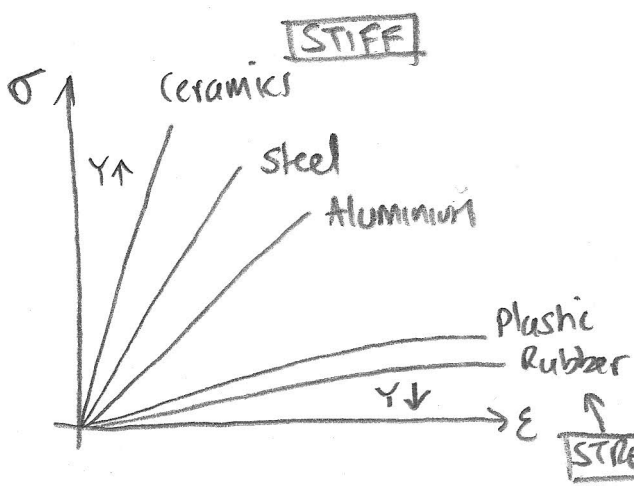
- σ_y Yield stress - lowest stress for permanent deformation
- σ_u Ultimate stress - stress which leads to DUCTILE or BRITTLE failure

- Y is constant during Hookean region.
- when load is removed in elastic region material returns to original shape
- in plastic deformation region, material is permanently deformed.

FRACTURE results from the propagation of cracks. The latter originate from **DISLOCATIONS** in the atomic structure, i.e. a slip of one plane of atoms over another



NOT ALL MATERIALS YIELD e.g. CERAMIC



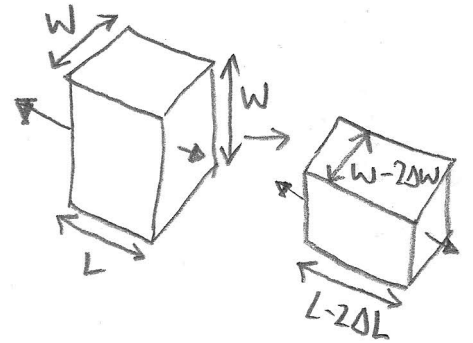
Stiff materials have a high Y often fail in a brittle manner.
 "Elastic" or **stretchy** materials have a low Y i.e. large strain ϵ for small stresses σ

Strong materials have a high yield (and ultimate) stress

Tough materials are somewhat resilient to brittle fracture. i.e. metals are often tough, but ceramics are not.

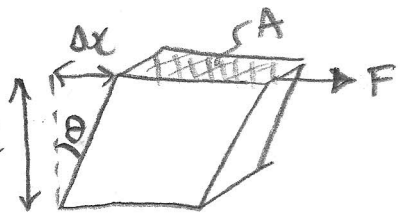
Most materials will shrink transversely as they are stretched. This is a positive **POISSON RATIO**

$$\nu = \frac{-\text{transverse strain change}}{\text{axial strain change}} \approx \frac{\Delta W}{\Delta L}$$



SHEAR MODULUS

$$G = \frac{\text{Shear stress}}{\text{Shear strain}} = \frac{FL}{A\Delta x}$$

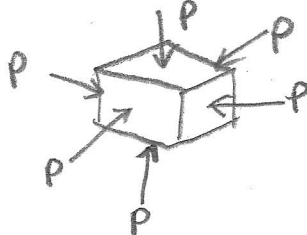


Note fluids and gases will flow rather than shear
 i.e. $G=0$

The compressibility of a material is described by the **BULK MODULUS**

$$K = -v \frac{dP}{dv} = \rho \frac{dP}{d\rho}$$

ρ is density
 P is pressure
 v is volume



i.e. ratio of a change in pressure P applied to result in the fractional change of volume \checkmark

NOTE FOR **ISOTROPIC** MATERIALS (i.e. equally strong in all directions - i.e. NOT like Carbon fibre!)

$$K = \frac{Y}{3(1-2\nu)} \quad Y = 2G(1+\nu)$$

For isotropic materials, the elastic moduli give the speed of sound

Pressure (P) waves

Shear (S) waves

$$c_p = \sqrt{\frac{k + \frac{4}{3}G}{\rho}}$$

$$c_s = \sqrt{\frac{G}{\rho}}$$

Example data: $[GPa = 10^9 N/m^2]$

MATERIAL	γ/GPa	ν	G/GPa	K/GPa	ρ/kgm^{-3}	c_p/ms^{-1}
RUBBER	0.01	0.5	0.0006	1	801	1120
STEEL	200	0.3	79.3	160	7800	5840
COPPER	117	0.3	44.7	123	8960	4510
PLASTIC	0.5 → 3	0.3 → 0.5	0.1	2.9	930	1810
CONCRETE	30	0.1 → 0.2	21	14.3	2400	4200
DIAMOND	1050 → 1210	0.107	478	443	3510	17540
WOOD	11	0.2 → 0.7	13	36.7	600-900	8490
GLASS	50 → 90	0.18 → 0.3	26.2	35 → 55	2500	5560

Young's modulus

Poisson ratio

shear modulus

Bulk modulus

Density

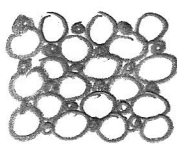
Pressure wave speed

Note speed of (P) waves in air $\approx 340 ms^{-1}$
 " " " " " water $\approx 1480 ms^{-1}$

Crystal structure is the (3D) pattern of ions/atoms/molecules in a material. A crystal usually implies an ordered, repeating, pattern of a distinct 'unit cell'

Different packing arrangements of atoms of same size/type

eg hexagonal packing



○ Sodium Na^+ ion
 ○ chloride Cl^- ion

← Repeating pattern of NaCl (Salt)